Two New Iridoid Esters from the Root and Rhizome of Valeriana jatamansi JONES

by Bo Yang^a)¹), Jian-Fen Zhang^b)¹), Hui-Zhu Song^a), Man-Cang Gu^a), Hua-Jun Zhao^{*a}), and Yao-Kang Xiong^{*a})

 a) College of Pharmaceutical Sciences, Zhejiang Chinese Medical University, Hangzhou 310053,
P. R. China (phone/fax: +86-571-86633145; e-mail: zhjzy@aliyun.com; phone/fax: +86-571-86633118; e-mail: xiongyaokang@126.com)

^b) College Biology and Environmental Engineering, Zhejiang Shuren University, Hangzhou 310015, P. R. China

Two new iridoid esters, named valerjatadoids A and B (1 and 2, resp.), together with three known iridoid esters, jatamanin O (3), jatamanvaltrate P (4), and jatamanvaltrate Q (5), have been isolated from the root and rhizome of *Valeriana jatamansi* JONEs. The structures of the two new compounds were elucidated by spectroscopic analyses, including 2D-NMR techniques.

Introduction. – The genus *Valeriana* (Valerianaceae), containing about 200 species, is distributed throughout the world. The root preparation of *Valeriana officinalis* L., which is commonly referred to as valerian, has been employed as a mild sedative in Europe for centuries [1]. *Valeriana jatamansi* JONES (Valeriana), is one of the crude drugs which has been used in China as a traditional folk medicine for treatment of various diseases including sleep disorder, nervous disorders, epilepsy, insanity, snake poisoning, and skin diseases [2][3]. Previous phytochemical studies on *V. jatamansi* revealed the presence of sequiterpenoids [4], essential oil [5], and a series of iridoid esters [6–8]. In our effort to search for new bioactive natural products from this plant, two new iridoid esters, yalerjatadoid A and valerjatadoid B (1 and 2, resp.), along with three known iridoid esters, jatamanin O (3), jatamanvaltrate P (4), and jatamanvaltrate Q (5) (*Fig. 1*), were isolated from the root and rhizome of *V. jatamansi*. Herein, we report the isolation and structure elucidation of the new compounds.

Results and Discussion. – Repeated normal phase and reversed phase column chromatography of the CH_2Cl_2 -soluble fraction of the 95% EtOH extract from the root and rhizome of *Valeriana jatamansi* JONES. (DEVJ) afforded compounds 1-5.

Compound **1** was obtained as yellow oil, its molecular formula $C_{24}H_{36}O_9$, was determined by HR-ESI-MS (491.2246 ($[M + Na]^+$, $C_{24}H_{36}NaO_9^+$; calc. 491.2257)). The IR spectrum indicated the presence of OH (3473 cm⁻¹) and ester CO groups (1738 cm⁻¹). From the ¹H-NMR spectrum, the presence of two isovaleroxy groups, one AcO group, and one EtO group was evident. The ¹³C-NMR spectrum showed 24 C-atom signals. In addition to the 14 C-atom signals for the substituents (two isovaleroxy,

¹) These authors have contributed equally to this work.

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Fig. 1. Chemical structures of 1-5

one AcO, and one EtO group), there were ten additional C-atoms composing the iridoid skeleton. The two C=C bonds in 1 were assigned to C(3)=C(4) (δ (H) 6.68 s, H–C(3); δ (C) 147.7 C(3) and 108.7 C(4)), and C(5)=C(6) (δ (H) 5.80 s, H–C(6); δ (C) 139.0 C(5) and 118.1 C(6)). After defining the iridoid skeleton and substituent moieties, an HMBC experiment was performed. The long-range correlations of Hatoms H–C(1), H–C(7), and CH₂(11) to C(1') at δ (C) 170.7, C(1'') at δ (C) 170.9, and $C(1^{\prime\prime\prime\prime})$ at $\delta(C)$ 171.9, respectively, of the acyloxy groups suggested that the two isovaleroxy groups and the AcO group were located at C(1), C(7), and C(11), respectively (Fig. 2). The EtO group was attached at C(10) based on the cross-peak of H-atoms CH₂(10) to the corresponding C-atom C(1''') at δ (C) 67.1 in the HMBC spectrum. By analyzing the HSQC, HMBC, and ¹H,¹H-COSY spectra, all the H- and Catom signals were assigned unambiguously. The relative configuration of 1 was elucidated by NOESY experiments. Since all the naturally occurring valepotriates exhibit an α -orientation for H–C(1) and β -orientation for H–C(9) [9][10], NOESY correlations were observed for $CH_2(10)/H-C(9)$, but not for $CH_2(10)/H-C(7)$ or H–C(9)/H–C(7), which implied that the isovaleroxy group at C(7) was β -oriented and the OH at C(8) α -oriented, which was consistent with the configurations of H–C(1), H-C(7), H-C(9), and $CH_2(10)$ of natural iridoids reported in the literature. Thus, compound 1 was elucidated as $(1R^*, 6R^*, 7S^*, 7aR^*)$ -4-[(acetyloxy)methyl]-7-(ethoxymethyl)-1,6,7,7a-tetrahydro-7-hydroxy-7a-methylcyclopenta[c]pyran-1,6-diyl bis(3methylbutanoate), and named valerjatadoid A.

Compound 2 was isolated as colorless oil and the molecular formula, $C_{32}H_{50}O_{12}$, was deduced from HR-ESI-MS (649.3189 ($[M + Na]^+$, $C_{32}H_{50}NaO_{12}^+$; calc. 649.3200)). The IR spectrum also indicated the presence of OH (3496 cm⁻¹) and ester CO groups (1740 cm⁻¹). Analysis of the ¹H- and ¹³C-NMR spectroscopic data (Tables 1 and 2) indicated signals for a trisubstituted olefinic bond (δ (H) 6.47 s, H–C(3); δ (C) 141.0 C(3) and 112.9 C(4)), and an acetal (δ (H) 6.24 d, J = 4.6, H–C(1); δ (C) 89.2 C(1)). These data, and the presence of five ester CO groups at δ (C) 173.0, 170.2, 169.7, 173.3, and 173.0, led to the conclusion that compound **2** is a 5,6-dihydrovaltrate iridoid with acyl substituents. The four acyl substituents were readily assigned by comparison with the ¹H- and ¹³C-NMR data from the reported valepotriates as two isovaleroxy, an AcO, and an α -(isovaleroxy) isovaleroxy group, respectively. The connectivities of the acyloxy substituents to the iridoid nucleus were fully assigned by a HMBC experiment. The linkage of an isovaleroxy group to C(1) was established by the HMBC from H–C(1) to the CO C-atom at δ (C) 173.0. The connection of the AcO group to C(7) was proposed on the basis of the HMBC from H–C(7) to the CO C-atom at δ (C) 170.2. Further, HMBC interactions between $CH_2(10)$ and the CO C-atom at $\delta(C)$ 169.7 led to the assignment of α -(isovaleroxy) isovaleroxy group to C(10), leaving an isovaleroxy residue to be located at C(11) (Fig. 2). HMBC interactions were also observed between

	Position	1	2
	1	6.29 (d, J = 10.0, 1 H)	6.24 (d, J = 4.6, 1 H)
	3	6.68 (s, 1 H)	6.47 (s, 1 H)
	5		2.91 (dd, J = 16.0, 7.8, 1 H)
	6	5.80 (s, 1 H)	2.12–2.14 (<i>m</i> , 2 H)
	7	5.47 $(d, J = 2.8, 1 \text{ H})$	5.01 (t, 4.5, 1 H)
	9	2.86 (dd, J = 10.0, 2.6, 1 H)	2.42 (dd, J = 9.5, 4.6, 1 H)
	10	3.57, 3.77 (2 d, J = 9.1, 2 H)	4.31, 4.37 (2 d, J = 11.5, 2 H)
	11	4.66, 4.72 (d, J = 12.4, 2 H)	4.44, 4.64 (2 d, J = 12.4, 2 H)
\mathbb{R}^1	2'	2.31–2.32 (<i>m</i> , 2 H)	2.20–2.22 (<i>m</i> , 2 H)
	3'	2.09–2.13 (<i>m</i> , 1 H)	2.13–2.14 (<i>m</i> , 1 H)
	4′	1.02 (d, J = 1.3, 3 H)	1.02 (d, J = 6.8, 6 H)
	5'	1.03 (d, J = 1.3, 3 H)	
R ⁷	2''	2.18–2.19 (<i>m</i> , 2 H)	2.09 (s, 3 H)
	3''	2.09–2.13 (<i>m</i> , 1 H)	
	4′′	0.97 (d, J = 2.3, 3 H)	
	5''	0.98 (d, J = 2.3, 3 H)	
R^{10}	1'''	3.53, 3.55 (2 d, J = 7.0, 2 H)	
	2'''	1.20 (t, J = 7.0, 3 H)	4.76 (d, J = 4.7, 1 H)
	3'''		1.28 (s, 1 H)
	4′′′		1.01 (s, 6 H)
	7'''		2.25 - 2.26 (m, 2 H)
	8'''		2.14 - 2.15 (m, 1 H)
	9′′′		1.00 (d, J = 2.8, 6 H)
R ¹¹	2''''	2.06 (s, 3 H)	2.31 - 2.32 (m, 2 H)
	3''''		2.14 - 2.15 (m, 1 H)
	4''''		0.98 (d, J = 6.6, 6 H)

Table 1. ¹*H*-*NMR Data of* **1** *and* **2** (at 600 MHz, in CDCl₃; δ in ppm, *J* in Hz)

1		
	92.7	89.2
3	147.7	141.0
4	108.7	112.9
5	139.0	31.8
6	118.1	34.9
7	83.7	80.1
8	80.2	80.9
9	48.1	44.5
10	71.1	66.8
11	61.1	63.4
1'	170.7	173.0
2'	43.3	43.5
3'	25.7	25.7
4′	22.3	22.4
5'	22.3	22.4
1″	170.9	170.2
2"	43.7	21.0
	25.8	
	22.4	
	22.5	
1‴	67.1	169.7
2′′′	15.0	76.9
3′′′		30.0
4′′′		18.7
5′′′		17.3
6'''		173.3
7'''		43.0
8'''		25.6
9′′′		22.3
10'''		22.3
1''''	171.9	173.0
2''''	20.9	43.0
3''''		25.7
4''''		22.4
5''''		22.4
	$\begin{array}{c} 4\\ 5\\ 6\\ 7\\ 8\\ 9\\ 10\\ 11\\ 1'\\ 2'\\ 3'\\ 4'\\ 5'\\ 1''\\ 2''\\ 3''\\ 4''\\ 5'\\ 1''\\ 2''\\ 3'''\\ 4'''\\ 5'''\\ 6'''\\ 7'''\\ 8'''\\ 9'''\\ 10'''\\ 1''''\\ 2''''\\ 3''''\\ 5'''\\ 5''\\ 5''''\\ 5''''\\ 5''''\\ 5''''\\ 5''''\\ 5''''\\ 5''''\\ 5''''\\ 5''''\\ 5''''\\ 5''''$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$

Table 2. ¹³C-NMR Data of **1** and **2** (at 150 MHz, in CDCl₃; δ in ppm, J in Hz)

CH₂(11) and the CO C-atom at δ (C) 173.0, in full support of this assignment. Furthermore, all of the key NOESY correlations supporting the structure of **1** were also observed in the 2D-NMR spectrum of **2**, suggesting that the relative configuration of **2** was the same as **1**. Accordingly, **2** was identified as [(1*R**,6*R**,7*S**,7a*R**)-6-(acetyloxy)-1,4a,5,6,7,7a-hexahydro-7-hydroxy-1-[(3-methylbutanoyl)oxy]-4-{[(3-methylbutanoyl)oxy]-uthyl}cyclopenta[c]pyran-7-yl]methyl 3-methyl-2-[(3-methylbutanoyl)oxy]-butanoate and named valerjatadoid B.

The known compounds were identified as jatamanin O (3) [11], jatamanvaltrate P (4) [12], and jatamanvaltrate Q (5) [12] by comparing their ¹H- and ¹³C-NMR data with those reported in the literature. All these three known compounds have been previously reported from *V. jatamansi*.

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Fig. 2. Key HMBCs (arrows), and COSY (bold lines), and NOESY (double-headed arrows) interactions of **1** and **2**

Experimental Part

General. Anal. TLC: silica-gel plates (SiO₂; Qingdao Ocean Chemical Co., Ltd.), with petroleum ether (PE)/AcOEt; detection by spraying with Epstahl reagent [13], followed by heating. Column chromatography (CC): silica gel (SiO₂; 100–200 mesh; Qingdao Ocean Chemical Co., Ltd.). Medium pressure chromatograph was carried out using a Büchi system consisting of C-605 pump and C-615 pump manager. ODS (40–60 µm, Daiso Co., Ltd., Japan). Semiprep. HPLC was carried out using a Waters system consisting of a 600 pump and a 2487 dual wavelength detector. The Amethyst C₁₈ reversed-phase column (250 × 21.2 mm, 10 µm) for semiprep. HPLC was purchased from Sepax Technologies (USA). Optical rotations: Autopol IV spectropolarimeter. IR Spectra: Bruker Tensor II spectrophotometer (Bruker, Germany), compounds were dissolved using CHCl₃ to form liquid film; in cm⁻¹. ¹H- and ¹³C-NMR spectra were measured on Bruker Avance 600 spectrometer, in CDCl₃; δ in ppm, J in Hz. Catom multiplicities were determined by DEPT-135 experiments. All 2D-NMR spectra were recorded using pulsed field gradients. One bond ¹³C,¹H-correlations were observed in an HSQC experiment. Longrange ¹³C,¹H-correlations were observed in HMBC. HR-ESI-MS: Thermo Scientific Exactive; in m/z.

Plant Material. The root and rhizome of *Valeriana jatamansi* JONES were collected in TongRen, Guizhou Province, P. R. China, in October of 2012, and identified by Professor *Ru-Song Zhang*, College of Pharmaceutical Sciences, Zhejiang Chinese Medical University, Zhejiang, P. R. China. A voucher specimen (No. ZYGC-VJ-20121001) was deposited with the Laboratory of TCM Resources Engineering, Zhejiang Chinese Medical University.

Extraction and Isolation. The dried rhizome and root of *V. jatamansi* (2.0 kg) were crushed into coarse powder, and then the powder was soaked with 95% EtOH for 24 h, and extracted with 20-fold 95% EtOH at r.t. by percolation. The combined extract was concentrated *in vacuo* to give 148.4 g of crude extract, and the extract was suspended in 1 l of warm H₂O (30°) and partitioned with CH₂Cl₂ (3×1 l). After evaporation under reduced pressure, the CH₂Cl₂ extract (98.5 g) was subjected to CC (SiO₂) eluting with a gradient of PE/AcOEt (40:1, 20:1, 15:1, 12:1, 10:1, 8:1, 6:1, 4:1, 2:1, 1:1, 0:1) to give eleven fractions (*Frs. A – K*). *Fr. G* (5 g of 6.8 g) was separated by prep. MPLC using MeOH/H₂O as the mobile phase (60%, 70%, 75%, 80%, 90%, each 21), flow rate 50 ml/min, to obtain compound **1**

(255 mg), compound **3** (199 mg), and compound **4** (169.8 mg). Some of the fractions eluted with 90% MeOH (305 mg) were subjected to prep. HPLC (82% MeOH/H₂O, flow rate 13 ml/min) to obtain compound **2** (64 mg, $t_{\rm R}$ = 42.2 min) and compound **5** (106 mg, $t_{\rm R}$ = 47.3 min).

Valerjatadoid A (=(1R*,6R*,7S*,7aR*)-4-[(*Acetyloxy*)*methyl*]-7-(*ethoxymethyl*)-1,6,7,7*a-tetrahy-dro-7-hydroxy-7a-methylcyclopenta*[c]*pyran-1*,6-*diyl Bis*(3-*methylbutanoate*); **1**). Yellow oil. [*a*]_{25}^{25} = 130.9 (*c* = 0.10, MeOH). IR (CHCl₃): 3473, 2961, 2873, 1738, 1642, 1487, 1370, 1243, 1101, 1023. ¹H-and ¹³C-NMR: see *Tables 1* and 2, resp. HR-ESI-MS: 491.2246 ([M + Na]⁺, C₂₄H₃₆NaO⁺₉; calc. 491.2257).

Valerjatadoid B (= [(1R*,6R*,7S*,7aR*)-6-(Acetyloxy)-1,4a,5,6,7,7a-hexahydro-7-hydroxy-1-[(3-methylbutanoyl)oxy]-4-[[(3-methylbutanoyl)oxy]methyl]cyclopenta[c]pyran-7-yl]methyl 3-Methyl-2-[(3-methylbutanoyl)oxy]butanoate; **2**). Colorless oil. $[a]_{D}^{25} = -2.4$ (c = 0.10, MeOH). IR (CHCl₃): 3496, 2962, 2874, 1740, 1467, 1372, 1327, 1032. ¹H- and ¹³C-NMR: see *Tables 1* and 2, resp. HR-ESI-MS: 649.3189 ([M + Na]⁺, $C_{32}H_{50}NaO_{12}^{+}$; calc. 649.3200).

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